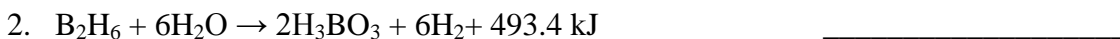


**ENDOTHERMIC VS. EXOTHERMIC WORKSHEET**

Identify the following reactions as endothermic or exothermic.



Thermochemical equations can be written 2 ways; with the heat written on the reactant side (endothermic) or product side (exothermic) or written to the right of the reaction as  $\Delta H$  with the appropriate sign to indicate positive (endothermic) or negative reaction (exothermic). For the following thermochemical reactions, rewrite the reaction the second way.

